

Intended Use

For the quantitative determination of iron in human serum.

Principle

The iron in serum is dissociated from its Fe (III) - transferrin complex by the addition of an acidic buffer containing hydroxylamine. This addition reduces the Fe (III) to Fe (II). The chromogenic agent, Ferene, forms a highly colored Fe (II) - complex that is measured photometrically at 560 nm.

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Teco Diagnostics

Test:

Iron Reagent Set (I593 - 100)

Number of Tests:

100 tests

Format:

Liquid

Method:

Colorimetric

Testing Procedure:

Manual

Storage Temperature:

15 - 30°C

Wavelength:

560 nm

Expected Values:

Iron, Total: $60 - 150 \mu g/dL$ Iron Saturation: 20 - 55%

It is recommended that each laboratory establish its own range of expected values.

Linearity:

500 μg/dL

Reagent Deterioration:

Appearances of turbidity, possible mold growth, or crystal formation that will not readily dissolve are signs of reagent deterioration.

Failure to obtain accurate results in the assay of control materials may indicate reagent deterioration.

Limitations of Procedure:

Certain drugs and other substances are known to influence circulating iron levels. Iron contained in hemoglobin does not react in this method; therefore, slight hemolysis will not interfere. However, gross hemolysis (pink or red specimens) will contribute to the absorbance measured at the wavelength used and should be avoided.

To make tubes, pipettes, etc. iron free, they must be washed with hot dilute 1:3 hydrochloric or nitric acid, followed by several rinsing with iron-free deionized or distilled water.